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(71) Applicant: **SHIN-ETSU CHEMICAL CO., LTD.**
Chiyoda-ku Tokyo (JP)

(72) Inventors:
• **Matsuda, Takashi**
Matsuida-machi, Usui-gun, Gunma-ken (JP)
• **Sato, Shinichi**
Matsuida-machi, Usui-gun, Gunma-ken (JP)

(74) Representative: **Stoner, Gerard Patrick et al**
MEWBURN ELLIS
York House
23 Kingsway
London WC2B 6HP (GB)

(54) Curable fluoropolyether base rubber compositions

(57) In a curable fluoropolyether base rubber composition comprising a linear fluoropolyether compound having at least two alkenyl groups, an organosilicon compound having at least two SiH groups, and a hydrosilylation catalyst, surface-hydrophobicized micro-

particulate silica having a specific surface area of at least 50 m²/g and a nitrogen atom content of 500-5,000 ppm is blended. A cured product thereof is significantly improved in compression set. The composition can be used in a wider variety of molding applications including O-rings.

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Description

[0001] This invention relates to curable fluoropolyether base rubber compositions capable of forming cured products having improved compression set as well as water repellency, oil repellency, solvent resistance, chemical resistance, weather resistance, and mold release and suitable for use as O-rings and the like.

[0002] A variety of microparticulate silica products surface treated to be hydrophobic are currently available. Most of them have a nitrogen atom content of less than 100 ppm and are of such nature that when 4 g of silica is dispersed in a mixture of 50 ml of pure water and 50 ml of methanol, the dispersion exhibits pH 4 to 7.

[0003] There are known compositions comprising a linear fluoropolyether compound having at least two alkenyl groups in a molecule and a perfluoroalkyl ether structure in its backbone, an organosilicon compound having at least two hydrogen atoms each attached to a silicon atom in a molecule, and a hydrosilylation catalyst. It was recently found that when microparticulate silica surface treated to be hydrophobic and having a surface area of at least 50 m²/g is blended in the compositions, the cured parts thereof are substantially improved in mechanical strength. The blending of silica gives fluoropolyether base rubber compositions having a good balance of many properties including heat resistance, chemical resistance, solvent resistance, mold release, water repellency, oil repellency, and weather resistance.

[0004] Although these fluoropolyether base rubber compositions exert satisfactory performance in most applications, they are unsatisfactory in molding applications to form O-rings and gaskets where a lower compression set is required. It would be desirable to improve the compression set of these compositions.

[0005] An object of the invention is to provide a novel and useful curable fluoropolyether base rubber composition capable of forming a cured product preferably having good compression set, combined with repellency, oil repellency, solvent resistance, chemical resistance, weather resistance, and mold release.

[0006] The invention is directed to a curable fluoropolyether base rubber composition comprising a linear fluoropolyether compound having at least two alkenyl groups in a molecule and a perfluoroalkyl ether structure in its backbone, an organosilicon compound having at least two hydrogen atoms each attached to a silicon atom in a molecule, and a hydrosilylation catalyst. The inventor has found that when a specific amount of surface-hydrophobized microparticulate silica having a specific surface area of at least 50 m²/g and a nitrogen atom content of 500 to 5,000 ppm is blended in this composition, the resulting cured product exhibited not only the properties including heat resistance, chemical resistance, solvent resistance, mold release, water repellency, oil repellency, and weather resistance characteristic of the known fluoropolyether rubbers incorporating the standard surface hydrophobized microparticulate silica, but was significantly improved in compression set. Then the composition can be used in a wider variety of molding applications including O-rings and gaskets.

[0007] The invention provides a curable fluoropolyether base rubber composition comprising (A) 100 parts by weight of a linear fluoropolyether compound having at least two alkenyl groups in a molecule and a perfluoroalkyl ether structure in its backbone, (B) 0.1 to 50 parts by weight of an organosilicon compound having at least two hydrogen atoms each attached to a silicon atom in a molecule, (C) a catalytic amount of a hydrosilylation catalyst, and (D) 5 to 60 parts by weight of microparticulate silica surface treated to be hydrophobic and having a specific surface area of at least 50 m²/g and a nitrogen atom content of 500 to 5,000 ppm.

[0008] Preferably the hydrophobized microparticulate silica (D) is such that when 4 g of the silica is dispersed in a mixture of 50 ml of pure water and 50 ml of methanol, the dispersion exhibits pH 8.5 to 10.5.

DESCRIPTION OF THE PREFERRED EMBODIMENTS

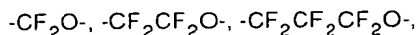
[0009] In the curable fluoropolyether base rubber composition of the invention, component (A) is a linear fluoropolyether compound having at least two alkenyl groups in a molecule and a perfluoroalkyl ether structure in its backbone.

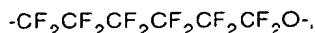
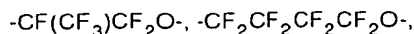
[0010] The perfluoroalkyl ether structure is represented, for example, by the following general formula (3):



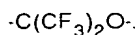
wherein Rf is a straight or branched perfluoroalkylene group of 1 to 6 carbon atoms, preferably 1 to 3 carbon atoms, and letter q is an integer of 1 to 500, preferably 2 to 400, more preferably 10 to 200.

[0011] Examples of the recurring units (Rf-O) of formula (3) are shown below.



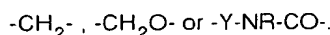


and

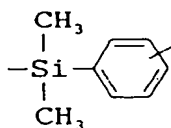


Of these, $-\text{CF}_2\text{O}-$, $-\text{CF}_2\text{CF}_2\text{O}-$, $-\text{CF}_2\text{CF}_2\text{CF}_2\text{O}-$, and $-\text{CF}(\text{CF}_3)\text{CF}_2\text{O}-$ are preferred. It is understood that the perfluoroalkyl ether structure may consist of recurring units of one type or recurring units of two or more types.

[0012] The alkenyl groups included in the linear fluoropolyether compound (A) are preferably those of 2 to 8 carbon atoms, especially 2 to 6 carbon atoms, and terminated with a $\text{CH}_2=\text{CH}-$ structure. Examples are vinyl, allyl, propenyl, isopropenyl, butenyl, and hexenyl, with vinyl and allyl being most preferred. The alkenyl groups may be positioned intermediate the molecule, but preferably attached to both ends of the molecular chain. In the latter case, the alkenyl groups may be attached either directly to both ends of the backbone of the linear fluoropolyether compound or to the backbone through a divalent linking group such as



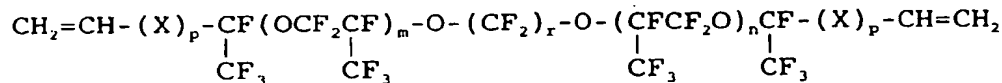
Herein Y is $-\text{CH}_2-$ or a group of the following structural formula (A):



(A)

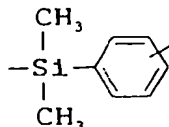
(the bond may be at o, m or p-position) and R is hydrogen, methyl, phenyl or allyl.

[0013] Of the linear fluoropolyether compounds (A) mentioned above, those of the following general formula (1) are preferred.



(1)

In formula (1), X is independently selected from among $-\text{CH}_2-$, $-\text{CH}_2\text{O}-$ and $-\text{Y}-\text{NR}-\text{CO}-$, letter p is independently equal to 0 or 1, r is an integer of 2 to 6, and m and n are integers of 0 to 200, preferably 5 to 100. Y is $-\text{CH}_2-$ or a group of the following structural formula (A):



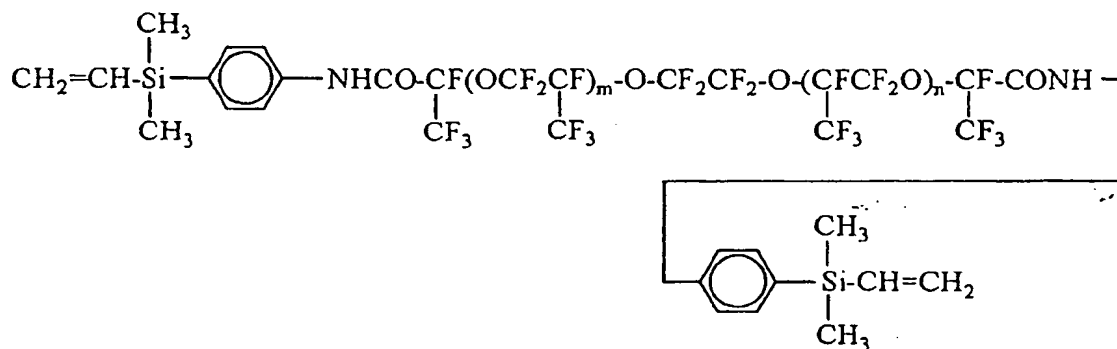
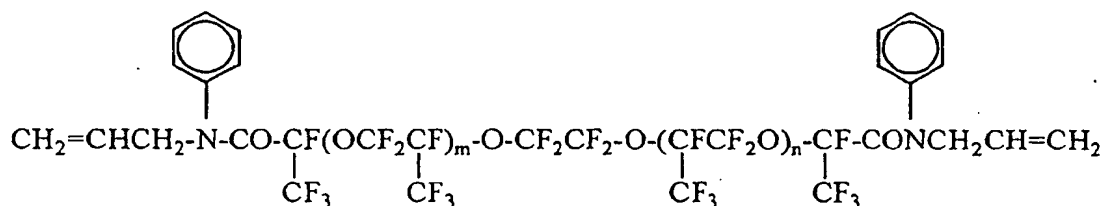
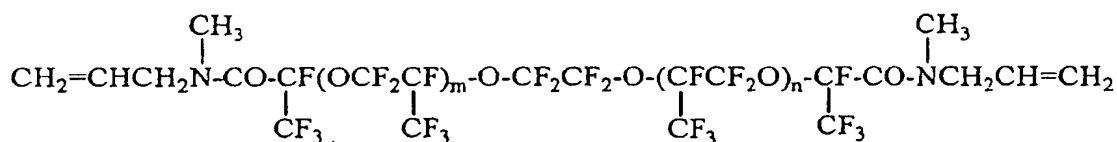
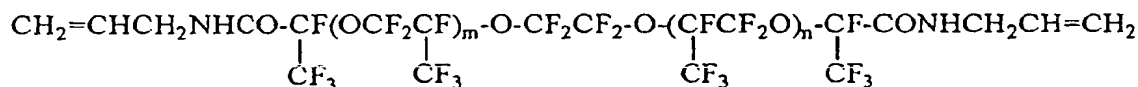
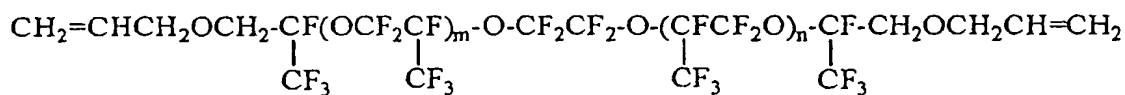
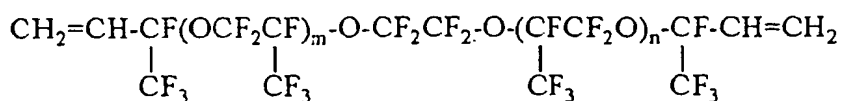
(A)

(the bond may be at o, m or p-position). R is hydrogen, methyl, phenyl or allyl.

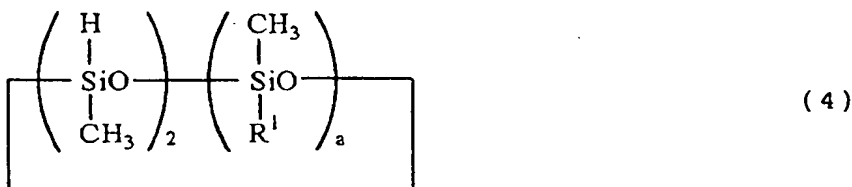
[0014] The linear fluoropolyether compound of formula (1) desirably has a weight average molecular weight of about

400 to about 100,000, and especially about 1,000 to about 50,000.

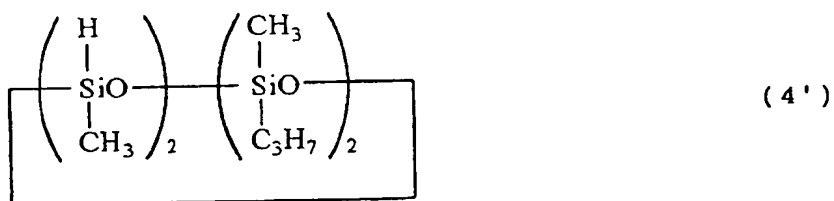
[0015] Illustrative examples of the fluoropolyether compound of formula (1) are given below. In the following formulae, m and n are as defined above.



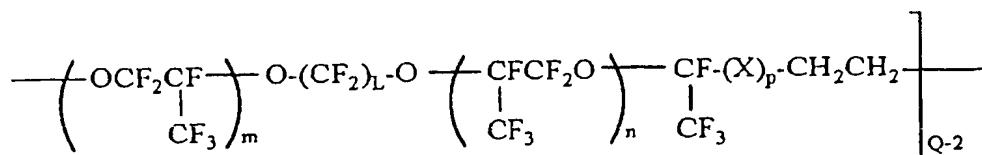
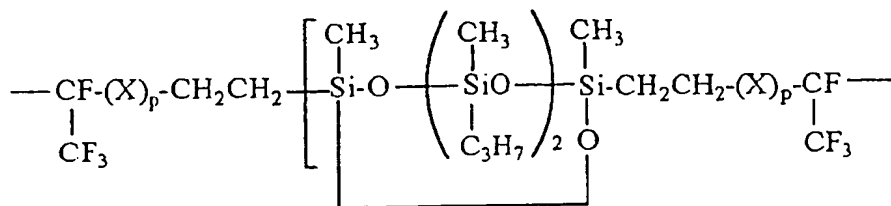
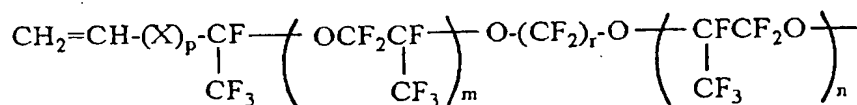
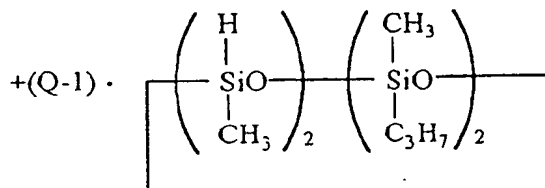
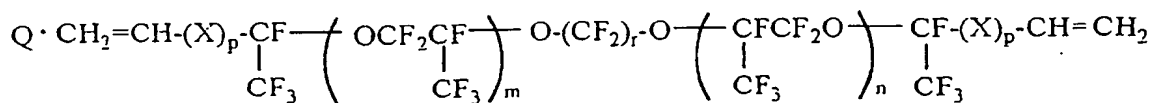
[0017] The organosilicon compound having two SiH groups in a molecule used to this end is preferably a straight or cyclic organohydrogenpolysiloxane of the following formula (4) or (5).

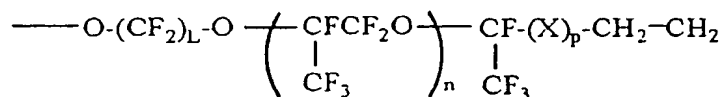
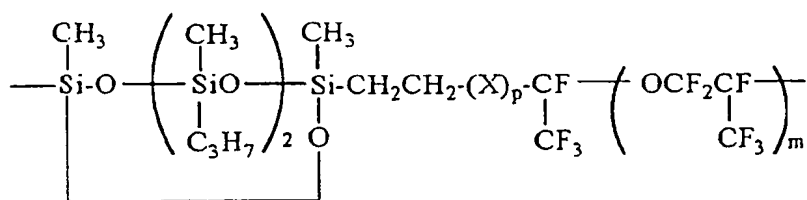

$$\begin{array}{c} \text{CH}_3 \\ | \\ \text{H}-\text{Si}-\text{O}-\left(\text{Si}-\text{O}\right)_b-\text{Si}-\text{H} \\ | \qquad | \qquad | \\ \text{R}^2 \quad \text{R}^2 \quad \text{CH}_3 \\ | \\ \text{CH}_3 \end{array} \quad (5)$$

[0018] Of these, the organohydrogenpolysiloxane of the following formula (4') is preferred.



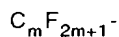
[0019] The organosilicon compound having two SiH groups in a molecule is reacted with the fluoropolyether compound of formula (1) under such conditions as to reach a desired extent of chain lengthening. The hydrosilylation reaction shown by the following reaction scheme gives a reaction product which can be used as component (A). It is noted that p, m, r, and n are as defined above and Q represents the moles of the reactant.



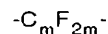


[0020] Component (B) is an organosilicon compound having at least two hydrogen atoms each attached to a silicon atom in a molecule, which functions as a crosslinking agent and chain extender for component (A). Any of organosilicon compounds having at least two hydrosilyl (SiH) groups in a molecule is useful. With the compatibility with and dispersibility in component (A), and uniformity after curing taken into account, organosilicon compounds having at least one monovalent perfluoroalkyl group, monovalent perfluorooxyalkyl group, divalent perfluoroalkylene group or divalent perfluorooxyalkylene group as well as at least two, preferably at least three hydrosilyl groups (or SiH groups) are preferred.

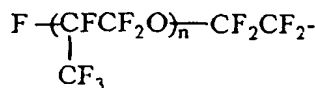
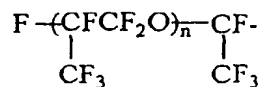
[0021] The perfluoroalkyl, perfluorooxyalkyl, perfluoroalkylene and perfluorooxyalkylene groups are exemplified by the groups of the following general formulac. monovalent perfluoroalkyl groups:



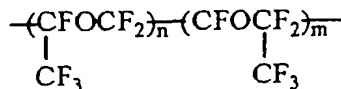
m is an integer of 1 to 20, preferably 2 to 10. divalent perfluoroalkylene groups:



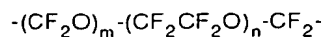
m is an integer of 1 to 20, preferably 2 to 10. monovalent perfluorooxyalkyl groups:



n is an integer of 1 to 5. divalent perfluorooxyalkylene groups:

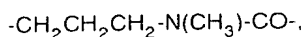
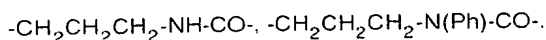
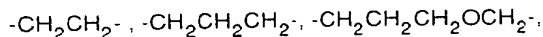


m+n is an integer of 2 to 100.

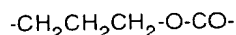


m and n each are an integer of 1 to 50.

[0022] These perfluoro(oxy)alkyl and perfluoro(oxy)alkylene groups each may be attached either directly to a silicon atom or to a silicon atom through a divalent linking group. The divalent linking group is an alkylene group, arylene group or a mixture thereof, which may further have an ether bond oxygen atom, amide bond or carbonyl bond. Such divalent linking groups of 2 to 12 carbon atoms are preferred. Illustrative examples thereof include



and



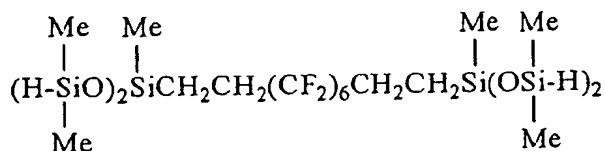
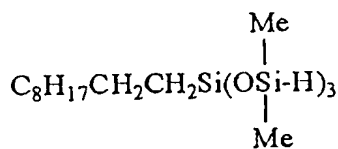
wherein Ph is phenyl.

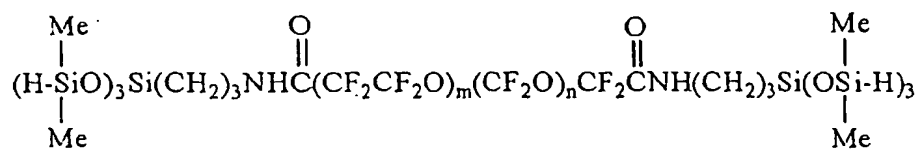
[0023] In addition to the monovalent organic group containing a monovalent or divalent fluorinated substituent, that is, a perfluoroalkyl, perfluorooxyalkyl, perfluoroalkylene or perfluorooxyalkylene group, the organosilicon compound (B) may have a monovalent substituent attached to a silicon atom. Exemplary monovalent substituents are substituted or unsubstituted hydrocarbon groups of 1 to 20 carbon atoms including alkyl groups such as methyl, ethyl, propyl, butyl, hexyl, cyclohexyl, octyl and decyl; alkenyl groups such as vinyl and allyl; aryl groups such as phenyl, tolyl, and naphthyl; aralkyl groups such as benzyl and phenylethyl; and substituted ones of these groups in which some of the hydrogen atoms are replaced by chlorine atoms, cyano groups or the like, such as chloromethyl, chloropropyl, and cyanoethyl.

[0024] The organosilicon compound (B) may be cyclic, chainlike or three-dimensional network.

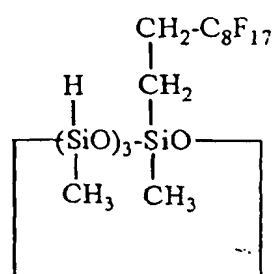
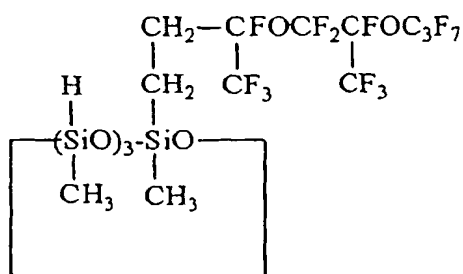
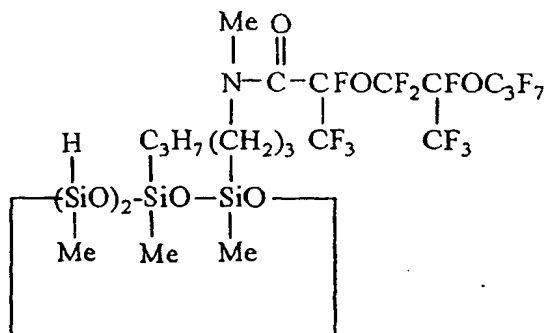
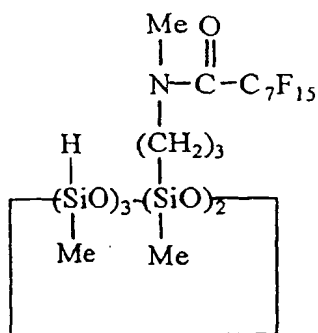
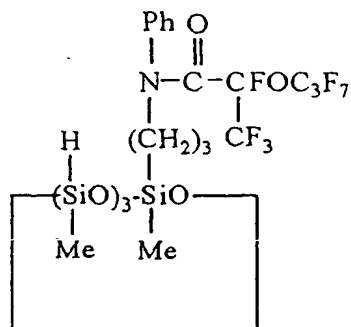
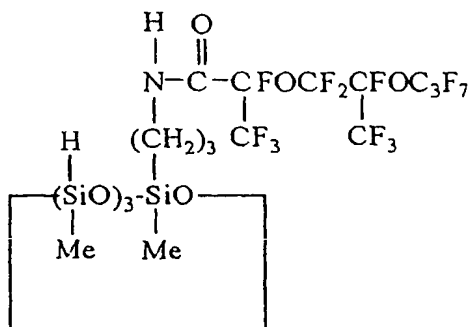
[0025] The number of silicon atoms in the molecule of the organosilicon compound is not critical although it desirably has about 2 to about 60 silicon atoms, and especially about 3 to about 30 silicon atoms.

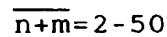
[0026] Illustrative examples of the organosilicon compound are given below. They may be used alone or in admixture of two or more. In the formulae, Me is methyl and Ph is phenyl.

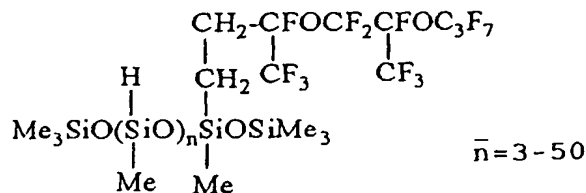
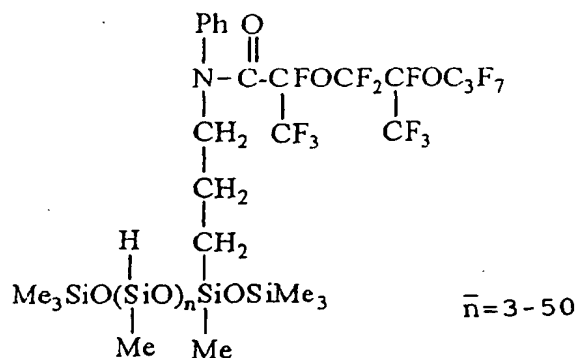
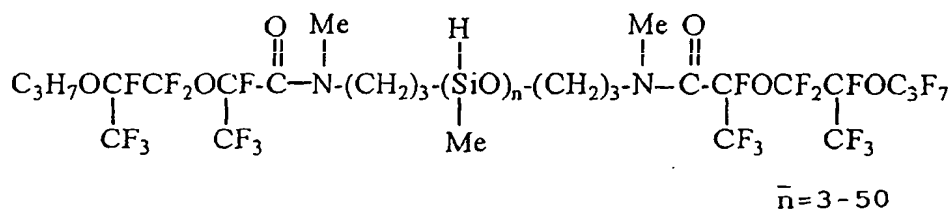
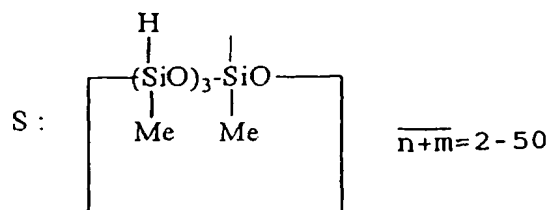
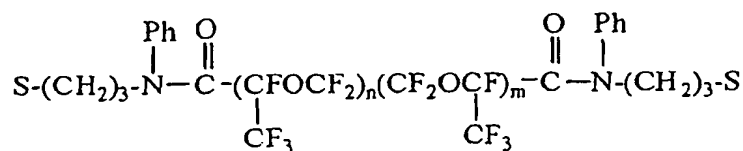




$$\bar{m}=10, \bar{n}=6$$







[0027] An appropriate amount of component (B) blended is 0.1 to 50 parts, and preferably 0.5 to 30 parts by weight per 100 parts by weight of component (A). Further preferably, component (B) having hydrosilyl groups is blended in such an amount that 0.5 to 5 mol, and more preferably 1 to 2 mol of the hydrosilyl (or SiH) groups may be present per mol of the alkenyl groups (e.g., vinyl, allyl or cycloalkenyl) in component (A). Less amounts of component (B) may achieve an insufficient degree of crosslinking. Excessive amounts of component (B) may allow chain lengthening to become preferential, inviting short curing, foaming, and losses of heat resistance and compression set.

[0028] Component (C) is a hydrosilylation catalyst which is preferably selected from transition metals, for example,

platinum group metals such as Pt, Rh and Pd, and compounds of transition metals. Most of these compounds are noble metal compounds which are expensive. Platinum compounds are thus used because they are readily available. Exemplary platinum compounds include chloroplatinic acid, complexes of chloroplatinic acid with olefins such as ethylene, complexes of chloroplatinic acid with alcohols and vinylsiloxanes, and platinum supported on silica, alumina or carbon though not limited thereto. Known platinum group metal compounds other than the platinum compounds include rhodium, ruthenium, iridium, and palladium compounds, for example, $\text{RhCl}(\text{PPh}_3)_3$, $\text{RhCl}(\text{CO})(\text{PPh}_3)_2$, $\text{RhCl}(\text{C}_2\text{H}_4)_2$, $\text{Ru}_3(\text{CO})_{12}$, $\text{IrCl}(\text{CO})(\text{PPh}_3)_2$, and $\text{Pd}(\text{PPh}_3)_4$ wherein Ph denotes phenyl.

[0029] The amount of the catalyst used is not critical. A catalytic amount can achieve a desired curing rate. From the economical aspect and to obtain satisfactory cured products, the platinum group metal compound is preferably added in an amount of 0.1 to 1,000 parts, more preferably 0.1 to 500 parts by weight calculated as the platinum group metal per million parts by weight of the entire curable composition.

[0030] Component (D) is microparticulate silica which has been surface treated to be hydrophobic and which has a specific surface area of at least $50 \text{ m}^2/\text{g}$ and preferably at least $80 \text{ m}^2/\text{g}$ (as measured by the BET method) and a nitrogen atom content of 500 to 5,000 ppm.

[0031] In general, microparticulate silica fillers are prepared by flame hydrolysis of silicon compounds such as silicon tetrachloride and methyltrichlorosilane. Microparticulate silica prepared by this method is useful in the practice of the invention. The microparticulate silica prepared by this method is hydrophilic because a plurality of silanol groups are present on the surface. Thus the silica must be hydrophobized.

[0032] Microparticulate silica fillers can be hydrophobized by well-known treatments, for example, by adding aliphatic alcohols to the silanol groups (USP 2,657,149); by contacting organochlorosilane vapor to convert the silanol groups to organosiloxy groups; by reacting organosiloxanes (JP-B 38-22129); and by reacting silazane compounds to convert the silanol groups to organosiloxy groups (USP 789,352).

[0033] Although the microparticulate silica used herein may be hydrophobized by any of these treatments, microparticulate silica treated with silazane compounds is preferred in the practice of the invention. Especially, microparticulate silica treated with cyclic silazane compounds of the following general formula (2) is preferred.



In formula (2), R is selected from halo-substituted or unsubstituted hydrocarbon groups of 1 to 6 carbon atoms including alkyl groups such as methyl, ethyl, propyl, butyl, hexyl, and cyclohexyl, alkenyl groups such as vinyl and allyl, phenyl, and substituted ones of these groups in which some of the hydrogen atoms are replaced by halogen atoms, such as chloromethyl, chloropropyl, 3,3,3-trifluoropropyl, and 3,3,4,4,5,5,6,6,6-nonafluorohexyl; and k is an integer of 3 to 10.

[0034] Of the cyclic silazane compounds of formula (2), the compound of the following general formula (2') is most preferred.



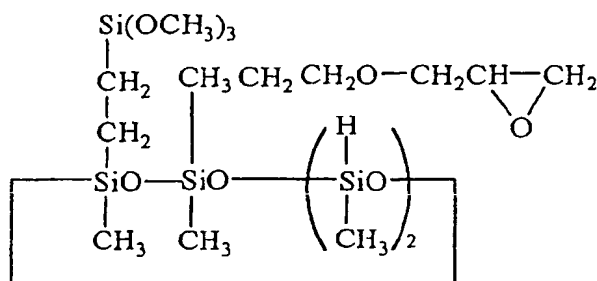
R is as defined above.

[0035] The microparticulate silica (D) should have a nitrogen atom content of at least 500 ppm. Maximum nitrogen content is preferably 5000 ppm, more preferably 2500 ppm. Lower nitrogen atom contents fail to provide satisfactory compression set. Excessive nitrogen atom contents may retard the catalysis of the hydrosilylation catalyst. The nitrogen content in microparticulate silica can be determined by Kjeldahl method, for example.

[0036] Further preferably, the microparticulate silica (D) is such that when 4 g of the silica is dispersed in a mixture of 50 ml of pure water and 50 ml of methanol, the dispersion exhibits pH 8.5 to 10.5 and especially pH 9.0 to 10.5. Silica giving lower pH may fail to achieve satisfactory compression set whereas silica giving higher pH can retard the catalysis of the hydrosilylation catalyst.

[0037] An appropriate amount of component (D) added is 5 to 60 parts, and preferably 10 to 40 parts by weight per 100 parts by weight of component (A). Less amounts of component (D) fail to achieve sufficient reinforcement. With excessive amounts of component (D) used, cured rubber is often affected thereby, for example, losing strength and elongation.

[0038] Insofar as the benefits of the invention are not impaired, various additives may be added to the inventive composition in addition to the above essential components (A) to (D). Such optional additives include inhibitors of the hydrosilylation catalyst, for example, acetylene alcohols such as 1-ethyl-1-hydroxycyclohexane, 3-methyl-1-butyn-3-ol, 3,5-dimethyl-1-hexyn-3-ol, 3-methyl-1-penten-3-ol, and phenylbutynol, as well as 3-methyl-3-penten-1-yn and 3,5-dimethyl-3-hexen-1-yn; tackifiers, for example, organosiloxanes having alkoxy, epoxy or SiH groups in the molecule such as the compound shown below; pigments such as iron oxide, cerium oxide and carbon black; colorants, dyes, and antioxidants. These are well-known.



[0039] Depending on its application, the rubber composition of the invention is formulated as a single composition having all the essential components (A) to (D) incorporated therein, that is, of one part type. Alternatively, the rubber composition is formulated to two part type, for example, one part containing components (A), (B) and (C) and the other part containing components (A), (B) and (D) whereupon the two parts are mixed on use.

[0040] The composition of the invention will cure when it is allowed to stand at room temperature or by heating. Often, the composition is preferably cured by heating at a temperature from room temperature (e.g., 10-30°C) to about 180°C for about 5 minutes to about 24 hours.

[0041] The curable fluoropolyether base rubber compositions cure into products having significantly improved compression set as well as heat resistance, chemical resistance, solvent resistance, mold release, water repellency, oil repellency, and weather resistance. Thus the compositions are useful in a wider variety of applications and suitable for use as O-rings and gaskets.

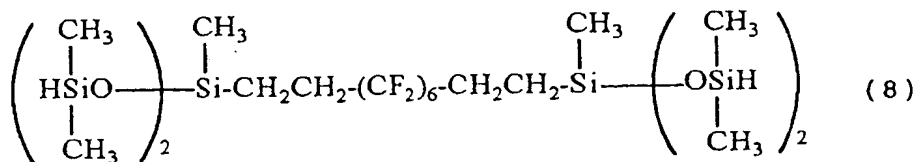
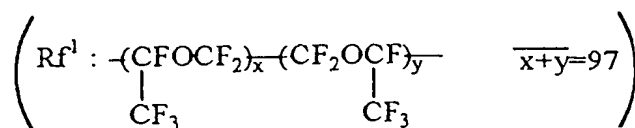
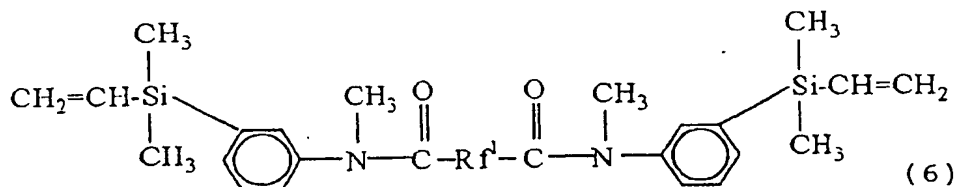
EXAMPLE

[0042] Examples of the invention are given below by way of illustration and not by way of limitation. All parts are by weight.

[0043] In the examples, the pH of hydrophobized silica was a measurement of a dispersion of 4 g of the silica in a mixture of 50 ml pure water and 50 ml methanol after it was agitated at 20°C for 30 minutes.

Example 1

[0044] To 100 parts of a polymer of formula (6) shown below (viscosity 4.4 Pa-s, average molecular weight 16500, and vinyl content 0.013 mol/100 g) was added 20 parts of hexamethylcyclotrisilazane-treated microparticulate silica (specific surface area 190 m²/g, surface silanol group number 0.43 group/nm², nitrogen atom content 2189 ppm, and pH 9.8). They were mixed and heat treated at 150°C for 2 hours, then milled on a three-roll mill. There were added 0.3 part of a 50% toluene solution of ethylcyclohexanol, 0.2 part of a toluene solution of a catalyst in the form of chloroplatinic acid modified with a compound of formula (7) shown below (platinum concentration 0.5% by weight), and 2.72 parts of an organosilicon compound containing SiH groups of formula (8) shown below. The ingredients were uniformly mixed, yielding a curable fluoropolyether base rubber composition.



[0045] From the composition, a cured sheet of 2 mm thick was prepared by press curing at 150°C for 5 minutes and oven curing at 200°C for 4 hours, and a set of balls for compression set measurement according to JIS K-6301 were prepared. The physical properties and compression set of these test pieces were measured according to JIS K-6301. The results are shown in Table 2.

Examples 2-3 & Comparative Examples 1-2

[0046] Curable fluoropolyether base rubber compositions and cured products were prepared by the same formulation and procedure as in Example 1 except that surface-hydrophobicized microparticulate silica as shown in Table 1 was used. The physical properties and compression set of the test pieces were measured, with the results shown in Table 2.

Table 1

EX2	microparticulate silica treated with hexamethylcyclotrisilazane (specific surface area 95 m ² /g, nitrogen atom content 1171 ppm, and pH 10.1)
EX3	microparticulate silica treated with hexamethyldisilazane (specific surface area 105 m ² /g, nitrogen atom content 877 ppm, and pH 9.2)
CE1	microparticulate silica treated with dichlorodimethylsilane (specific surface area 180 m ² /g, nitrogen atom content 20 ppm, and pH 4.8)
CE2	microparticulate silica treated with hexamethyldisilazane (specific surface area 110 m ² /g, nitrogen atom content 67 ppm, and pH 5.6)

Table 2

	EX1	EX2	EX3	CE1	CE2
Hardness (JIS-A)	64	59	57	62	56
Tensile strength (kgf/cm ²)	86	75	72	85	76
Tear strength (kgf/cm)	18	15	15	19	17
Elongation (%)	250	240	250	270	290
Compression set* (%)	12	13	18	30	26

* 200°C/24 hours

[0047] It is evident from Table 2 that as compared with cured products of comparative compositions having blended therein microparticulate silica whose nitrogen atom content or pH is outside the scope of the invention, cure products of curable fluoropolyether base rubber compositions within the scope of the invention have substantially equivalent physical properties, but are significantly improved in compression set. In particular, the cured products of Examples 1 and 2 using microparticulate silica surface treated with cyclic silazane compounds for hydrophobicization are very low in compression set variation.

[0048] Japanese Patent Application No. 11-053543 is incorporated herein by reference.

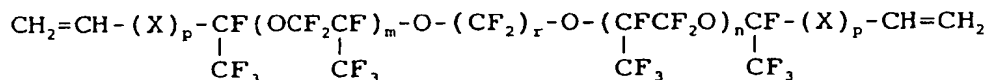
[0049] Although some preferred embodiments have been described, many modifications and variations may be made thereto in light of the above teachings. It is therefore to be understood that the invention may be practiced otherwise than as specifically described in the examples.

Claims

1. A curable fluoropolyether base rubber composition comprising:

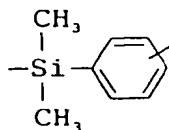
- (A) 100 parts by weight of linear fluoropolyether compound having at least two alkenyl groups per molecule and a perfluoroalkyl ether structure in its backbone;
- (B) 0.1 to 50 parts by weight of organosilicon compound having per molecule at least two Si-bonded hydrogen atoms;
- (C) a catalytic amount of hydrosilylation catalyst, and
- (D) 5 to 60 parts by weight of microparticulate silica having a hydrophobic surface treatment, a specific surface area of at least 50 m²/g and a nitrogen atom content of 500 to 5,000 ppm.

2. The composition of claim 1 wherein the linear fluoropolyether compound (A) is of the following general formula (1):



(1)

wherein X is independently -CH₂-, -CH₂O- or -Y-NR-CO- wherein Y is -CH₂- or a group of the following structural formula (A):



(A)

and R is hydrogen, methyl, phenyl or allyl,

letters p are independently equal to 0 or 1, r is an integer of 2 to 6, and m and n are integers of 0 to 200.

3. A composition of claim 1 or 2 wherein the microparticulate silica (D) has been surface treated for hydrophobicization with a cyclic silazane compound of the following general formula (2):



wherein R is independently a substituted or unsubstituted monovalent hydrocarbon group of 1 to 6 carbon atoms in which one or more hydrogen atoms may be replaced by halogen atoms, and k is an integer of 3 to 10.

4. The composition of claim 3 wherein the cyclic silazane compound of formula (2) is represented by the following general formula (2'):



wherein R is as defined above.

5. A composition according to any one of the preceding claims wherein said microparticulate silica is such that when 4 g of the silica is dispersed in a mixture of 50 ml of pure water and 50 ml of methanol, the dispersion exhibits pH 8.5 to 10.5.
6. An article comprising a cured rubber composition according to any one of the preceding claims.
7. An article according to claim 6 which is a seal element such as an O-ring or gasket.
8. In a composition as defined in any one of claims 1 to 5, the use of the nitrogen content of at least 500 ppm in the microparticulate silica to reduce the compression set of the cured composition.

(19)



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(71) Applicant: **SHIN-ETSU CHEMICAL CO., LTD.**
Chiyoda-ku Tokyo (JP)

(72) Inventors:
• **Matsuda, Takashi**
Matsuida-machi, Usui-gun, Gunma-ken (JP)
• **Sato, Shinichi**
Matsuida-machi, Usui-gun, Gunma-ken (JP)

(74) Representative: **Stoner, Gerard Patrick et al**
MEWBURN ELLIS York House 23 Kingsway
London WC2B 6HP (GB)

(54) Curable fluoropolyether base rubber compositions

(57) In a curable fluoropolyether base rubber composition comprising a linear fluoropolyether compound having at least two alkenyl groups, an organosilicon compound having at least two SiH groups, and a hydrosilylation catalyst, surface-hydrophobicized micro-particulate silica having a specific surface area of at

least 50 m²/g and a nitrogen atom content of 500-5,000 ppm is blended. A cured product thereof is significantly improved in compression set. The composition can be used in a wider variety of molding applications including O-rings.

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EUROPEAN SEARCH REPORT

Application Number
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DOCUMENTS CONSIDERED TO BE RELEVANT			
Category	Citation of document with indication where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int. Cl. 7)
X	US 5 837 774 A (FUKUDA KENICHI ET AL) 17 November 1998 (1998-11-17) * comparative example 1 * * column 1 - column 4; claims *	1,2,5-8	C08L71/02 C08G65/32 C08K3/36 C08G77/12 C08K9/06
Y	* comparative example 1 * * claim 1 *	1-8	
E	EP 1 114 846 A (SHINETSU CHEMICAL CO) 11 July 2001 (2001-07-11) * example 1; comparative example 1 * * claims *	1-8	
P,X	EP 0 967 251 A (SHINETSU CHEMICAL CO) 29 December 1999 (1999-12-29) * claims *	1-8	
X	EP 0 765 916 A (SHINETSU CHEMICAL CO) 2 April 1997 (1997-04-02) * example 1; example 2 * * claims *	1,2	
Y	US 5 591 797 A (BARTHEL HERBERT ET AL) 7 January 1997 (1997-01-07) * column 3 - column 5; claim 1 *	1-8	
Y	US 5 358 996 A (YAMAGUCHI KOUICHI ET AL) 25 October 1994 (1994-10-25) * claims; examples *	1-8	TECHNICAL FIELDS SEARCHED (Int. Cl. 7) C08K C08G
The present search report has been drawn up for all claims			
Place of search MUNICH		Date of completion of the search 12 October 2001	Examiner Sharma, U
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This annex lists the patent family members relating to the patent documents cited in the above-mentioned European search report. The members are as contained in the European Patent Office EDP file on
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12-10-2001

Patent document cited in search report		Publication date		Patent family member(s)	Publication date
US 5837774	A	17-11-1998	JP	3017657 B2	13-03-2000
			JP	8269317 A	15-10-1996
EP 1114846	A	11-07-2001	JP	2001192546 A	17-07-2001
			EP	1114846 A2	11-07-2001
			US	2001008916 A1	19-07-2001
EP 0967251	A	29-12-1999	JP	2000007835 A	11-01-2000
			EP	0967251 A1	29-12-1999
			US	6297339 B1	02-10-2001
EP 0765916	A	02-04-1997	JP	9095615 A	08-04-1997
			DE	69611023 D1	28-12-2000
			DE	69611023 T2	17-05-2001
			EP	0765916 A2	02-04-1997
			US	5656711 A	12-08-1997
US 5591797	A	07-01-1997	DE	4336345 A1	27-04-1995
			DE	4401598 A1	27-07-1995
			EP	0649885 A1	26-04-1995
			JP	2831581 B2	02-12-1998
			JP	7187648 A	25-07-1995
			KR	135217 B1	22-04-1998
US 5358996	A	25-10-1994	JP	6049348 A	22-02-1994

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For more details about this annex : see Official Journal of the European Patent Office, No. 12/82